

AP Biology 2000 Student Samples

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SECTION II

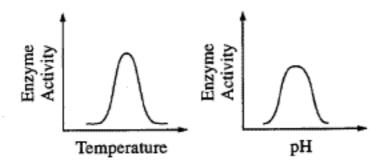
Time-90 minutes

4 Questions

Directions: Answer all questions. Write your answers on the pages following the questions in the pink booklet.

Answers must be in essay form. Outline form is NOT acceptable. Labeled diagrams may be used to supplement discussion, but in no case will a diagram alone suffice. It is important that you read each question completely before you begin to write.

 The effects of pH and temperature were studied for an enzyme-catalyzed reaction. The following results were obtained.



- a) How do (1) temperature and (2) pH affect the activity of this enzyme? In your answer, include a discussion of the relationship between the structure and the function of this enzyme, as well as a discussion of how structure and function of enzymes are affected by temperature and pH.
- b) Describe a controlled experiment that could have produced the data shown for either temperature or pH. Be sure to state the hypothesis that was tested here.

a. As the results from the experiment show, the enzyme has both an optimal temperature and an optimal ph. Deviation from this optimum results in a decrease in enzyme activity and efficiency. Enzymes are proteins whose unique structure allow them to catalyze chemical reactions. They have primary struct The sequence of amino acids, presense or absense of alpha helixes and beta

pleated Sheets, the types of bonds Cionic, covalent, disulfide, etc.) and the aggregation with an protein all effect an enzyme's ability to perform. This unique structure results in active site, an area that is conducive the reactants of a particular reaction. Each enzyme is very specific and can catalyze only one reaction. its active site one substrate (reactant). This only open to is is known as the lock and key theory of enzyme specifity. By temporarily hydrogen bonding to the substrate, the enzyme puts the reactions in positions which enhance the probability reaction occurring. When heat is applied to an enzyme, the movement of substrate increases and the substrates are more likely to come in contact with the enzyme's active site. This increases enzyme activity, however if much heat is applied, the enzyme will denature, its Shape will change, and the substrate will no longer be able to fit in the active site, decreasing enzyme activity. The same is true for pH. 每 If the pH is too acidic, the influx Ht may affect the bonds of the enzyme, as may an influx of OH if the pH is too basic, This will

also affect the shape of the enzyme's active site and adversely influence its ability to catalyse a reaction.

A controlled experiment could have been produced to show the effects of temperature on enzyme activity. The hypothesis would be that enzymes work best at warmer temperature, but if the temperature 0 too not, the enzyme will denature and its activity rate will decrease. In order to do this an enzyme and substrate must be Choosen. A good enzyme would be catalase, which increases the vate at which hydrogen peroxide (H2O2) breaks fown into water (H2O) and oxygen (O2). For the control, the enzyme and substrate would be placed in room temperature. Numerous beakers of enzyme and gubs trate would be allowed to sit for different time intervals, The reactions could be stopped by the addition of a strong acid to beakers. This would denature the enzymes conversion of H2O2 into H2O and and stop the Oz, A graph could then be profiled of enery in determine how active the enzyme was, a titration could be used. A titration the introduction of an indicator chemical

mat allows one to visualize the endpoint of
a reaction. A The amount of H20g remaining
could be determined and protted on a
graph. To check the affects of tempera
on enzyme activity, the experiment could
be run several times more, but each
time with a different temperature, some
colder and some warmer, By comparing
these various graphs to the graph of
the control, it would be possible to detern
the affects of heat on enzyme activity

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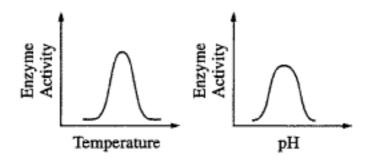
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The graphs Shown, indicate a mount shaped distribution
meaning that as temperature or pt poleveb increase, the enzyme
activity will also increase. This is reversed once a peak, maximum
potentials is reached. Then the further increase of temperature
or pt level will hinder Enymatic activitie. An enzyme
is produced to fit the new needs of one function. Each enzyme
can only bond with it respective substrate this model, often
called the lockard key model, allows for only slight variations in order
Thetwo match BB

The two match

OP at the binding

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each other

GO ON TO THE NEXT PAGE

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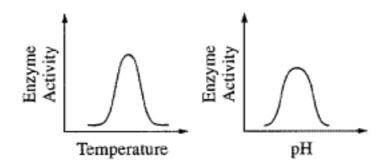
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Although they are unable to make a reaction occur that would not normally according hor they substantially lower the necessary activation energy for the reaction. This greatly increases the praction rate enzymes are also usually highly specific, which means that each engine catalyzes one or maybe a few chemical reactions.

There are two different theories as to how an enzyme forctions. The first is known as the lock-and-key model,

as shown below. Note that the structure of the entryme remains essentially unoringed. ock-and-key Model enzyme reactions enzyme L-bonding sites enzine-substrate complex Another theory is based on the induced- fit model. This is currently thought to be the most likely model of an ortune, as shown below. Note that the conformation of the protein changes stightly in the reactions reactions enzyme enzyme . enzyme-substrate Induced-Fit product. MODEL

Enzymes are affected by a variety of factors, including temperature and pH. Most enzymus record most effectively at between 20-30°C and (0-8 pH. When enzymos are exposed to temperature of pH extremes, it affects their tentiary bonding structure and may even permanently denatore the enzyme. However, some enzymus such as pepsin function at pH or temperature are extremes. Pepsins is most effective at low (acidic) pH's as are found in the Stomach.

b) The hypothesis is that enzymes are most effective at temperature
between 20° and 30°C. In order to test this an experiment
could be designed in which potatoes and the onzyme amyliase ene
prised-treater combined in a solutions of voting temperatury ranging
from 0°C to 50°C. This, there would be setups for 0°,5,10°, 520°,5
30%40 4 and 50°. Potatoes contain starch, and amplase is a engine
that breaks this acup. After a specified time period constant
for all temperatures the amount of product could be measured and
the data processed to detain a curve & graphed. This experiment
Should serve to reinforce the aforementationed hypothesis and produce
the following (UNE:
Temp
lenp

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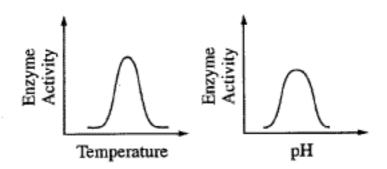
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All It seems to me that the enzyme is affected vell by temperature. There is a rise as the enzyme catalysis reaches its optimum temperature and then a full as temperatures get so hot that they actually inhibit catalysis. All As for pH it also has an optimum pH but it seems to be lover than the optimum temperature. The optimum pH is in the middle so it might be an

enzyme whose optimum pH is around 7 (not acid or
basic but neutral). As the pH scale rises
its enzyme activity increases but only until it
reaches the middle because when it does it
starts going down again as the environment
gets too basic.
B) My experiment would consist on testing for
the optimum temperature for the enzyme
the optimum temperature for the enzyme lacture on the catalyst lacture found in ordinary was milk.
in ordinary cows milk.
I would keep the milk in different beakers.
Exactly 5 beakers each filled with 500 ML of milk. Beaker #1 would be set at 55° F. Beaker #3
of milk. Beaker #1, would be at 40 F.
Beaker #2 would be set at 55 F. Beaker #3
vails be at Room Temperature \$ 70°F.
Beaker # 4 (0) 85 F. Beaker #5 @ 100 F.
I would grind up 16mes Lactaid Lactase captet
into each Beaker. Then led test for glacose
contest ineach in 5 min. The one with the
Wighest content wall be the optimin Temp!