

AP® Biology 2003 Sample Student Responses Form B

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- 3. Water is important for all living organisms. The functions of water are directly related to its physical properties.
 - (a) Describe how the properties of water contribute to TWO of the following.
 - · transpiration advises in -waller
 - thermoregulation in endotherms
 - plasma membrane structure
 - (b) Water serves as a reactant and a product in the carbon cycle. Discuss the role of water in the carbon cycle.
 - (c) Discuss the impact of one human activity on the water cycle.
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bodies after rains	a part of the water cycle and mix with the water
	bodies after rains.
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- 3. Water is important for all living organisms. The functions of water are directly related to its physical properties.
 - (a) **Describe** how the properties of water contribute to TWO of the following.
 - transpiration
 - thermoregulation in endotherms
 - plasma membrane structure
 - (b) Water serves as a reactant and a product in the carbon cycle. Discuss the role of water in the carbon cycle.
 - (c) Discuss the impact of one human activity on the water cycle.

a) the properties of water are essential for transpiration.
The hydrogen bonds between the oxogen of tryckrogen water
molecules gives nater the properties of whesion & adhesion. Water
moves up the xylem of plants by of diffusion. It moves from the areas u/ mest
water to areas of less water concentration. There, it diffuses into the air. The
water can move up the xylem in a column that does not break thanks
to its otherive properties- Each water molecule pulle the next one up,
etc. Adhesian to the sides of the xylem tokes of capillanty
of the liber also contribute in transpiration
' Di .
Plasma membrane structure
- Water it polar. The plasma membrane is
composed of a double phospholipid bilayer: - 999
: A double layer of phospholipids - Phospholipids - a & & &
phosphate group w/ 2 attached fatty acid tails are
amphiphatic molecules. The fatty acids are hydrophobic rotatestia
phosphate toados because they are non-polar. The phosphate head
is polar & hydrophilic Thus - these molecules analoge themselves
is polar of hydrophilic. Thus these molecules analoge themselves so that the hydrophilic region is no touch of the water of the lipid tails
are away from it. as so: 900
$\mathcal{O}(\mathcal{O}\mathcal{O})$ GO ON TO THE NEXT PAGE.

ADDITIONAL PAGE FOR ANSWERING QUESTION 3

b.) Hoter or polar to osphate grass, a how sold to
water is point of the raiben yell in photosynthesis
it is used, as a reactant along w/ CO2 in order to man produce
it is used as a reactant along w/ CO2 in order to man produce
qual- It is split by photolysis-d its électrons are excited by light
in the photosyspens tosof & used in the etransport chain- the Oz
released by plants is a byproduct that comes from splitting water.
In collular respiration, water is a product It is formed
when the O, posteros pulling down electrons from the
ē transport chain combines w/ & hydrogen.
<u> </u>
c) Human activity by polluting water resources
by the responsible for and rain.

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Water is a polar molecule, that is, it has a It & a I and The oxygen has a greater pull on The Shared electrons than does hydrogen because of its more paritive core. Oxygen has more protons than hydrogen does. Water molecules form hydrogen bonds to with other water molecules Hydrogen bonds are weak intermolecular bonds formed between the slightly dightly negatively charged atoms in this hydrogen otoms & case exygen. This attraction between water molecules is called cohesion Dater notecules are also attracted to other polar substances. This attraction is couled adhesion (chesion & adhesion are two major properties of water That contribute transpiration. Transpiration is water loss from the leaves of Stomata are small openings in the leaf through which water evaporates transpires. Water transpiring from the leaf "pull" on the it (whesien). Water thus moves up The xylem of the last water Adhesion helps the water by attracting The water incleances to The wall of The xylen

blater's high hear capacity contributes to Thermoregulation in endotherms. Hear capacity is the amount of energy needed to raise the temperature of 1g of substance by 1°C. Thermoregulation is the control of an organism's temperature to maintain homeostasis. An endotherm is an organism That does not rely on The environment to maintain its body temperature. Water is used by endotherms to lower body temperature. Humans are endoderms. When a human's

ADDITIONAL PAGE FOR ANSWERING QUESTION 3

body temperature goes above The desireable range, the body responds by
sweating. Sweat consists largely of water. The water absorbs heat from the
skin's surface & evaporates, removing hear from The skin & cooling The body
Water as a reactant in the carbon cycle is incorporated into glycose during
photogrammesic Water molecules contribute The electrons that are excited by
photons of light during the light-dependent stage of photosynthes. The
hydrogen atoms from water are Then incorporated into glucore
molecules during the light-independent process of the Calvin-Benier your
The ayon atoms from exygen gas & are released into The atmosphere. Water is
reproduced during cellular respiration when Oz act as an electron accepter at
The end of the electron transport chain & combines with H+ ions paring
Through ATP synthase in The mitochandria.
Tuman pollution, rawing an increase in atmospheric COz & a decrease
in ozone has caused by global warming. This increase in temperature
means That more water evaporates of transpines. This has chried up a
lot of land, creating more desert. The heat is also meeting The policy ice
capt country a rise in sea level. The global warming country pollution
includes activities such as using care & freen refrigerators.